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NEET CHEMICAL BONDING AND MOLECULAR STRUCTURE — FAJANS' RULE — PYQS

Max. Marks: 12 **Questions:** 3

Section A (MCQ) — 3 questions

Q.1 Which of the following compounds has the lowest melting point? [AIPMT 2011 (Prelims)]

(1) $CaCl_2$ (2) $CaBr_2$ (3) CaI_2 (4) CaF_2

Scan or tap for the full solution

Q.2 In the case of alkali metals, the covalent character decreases in the order [AIPMT 2009]

(1) MF > MCl > MBr > MI

(2) MF > MCl > MI > MBr

(3) MI > MBr > MCl > MF

(4) MCl > MI > MBr > MF

Scan or tap for the full solution

Q.3 The correct sequence of increasing covalent character is represented by [AIPMT 2005]

(1) LiCl < NaCl < $BeCl_2$

(2) $BeCl_2$ < LiCl < NaCl

(3) NaCl < LiCl < $BeCl_2$

(4) $BeCl_2$ < NaCl < LiCl

Scan or tap for the full solution

ANSWER KEY

Ques.	1	2	3								
Ans.	**3**	**3**	**3**								