

PART 1 - CHEMISTRY**Topic :** Classification of elements and periodicity in properties**SECTION-1 : (Maximum Marks : 4)**

- This section contains **ONE** question.
- Each question has **FOUR** options (A), (B), (C) and (D). **ONE OR MORE THAN ONE** of these four option(s) is(are) correct answer(s).
- For each question, choose the option(s) corresponding to (all) the correct answer(s).
- Answer to each question will be evaluated according to the following marking scheme :
Full Marks : +4 ONLY if (all) the correct option(s) is(are) chosen;
Partial Marks : +3 If all the four options are correct but ONLY three options are chosen;
Partial Marks : +2 If three or more options are correct but ONLY two options are chosen, both of which are correct;
Partial Marks : +1 If two or more options are correct but ONLY one option is chosen and it is a correct option;
Zero Marks : 0 If none of the options is chosen (i.e. the question is unanswered);
Negative Marks : -2 In all other cases.
- For example, in a question, if **(A), (B) and (D)** are the ONLY three options corresponding to correct answers, then choosing ONLY (A), (B) and (D) will get +4 marks; choosing ONLY (A) and (B) will get +2 marks; choosing ONLY (A) and (D) will get +2 marks; choosing ONLY (B) and (D) will get +2 marks; choosing ONLY (A) will get +1 mark; choosing ONLY (B) will get +1 mark; choosing ONLY (D) will get +1 mark; choosing no option (i.e. the question is unanswered) will get 0 marks; and choosing any other combination of options will get -2 marks.

1. The correct statement(s) regarding the periodic properties of elements is(are) [JEE Advanced 2026 Paper 2]
- (A) Second ionization enthalpy of carbon atom is less than that of boron atom.
- (B) Increasing order of ionic radii: $\text{Al}^{3+} < \text{Mg}^{2+} < \text{Na}^+$
- (C) Under identical conditions, in solid state, the density of potassium metal is more than density of sodium metal.
- (D) The H-H bond is weaker than F-F bond.

Scan or tap for the full solution

**Space for Rough Work**

JEE ADVANCED CLASSIFICATION OF ELEMENTS AND PERIODICITY IN PROPERTIES — PYQS

Test Pattern : **JEE (Advanced)**

ANSWER KEY

SECTION-1

[illegible]